



# Hematology



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## BIOCHEMSITRY

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Sheet

Slide

Handout

Number: **1**

Subject: **Hemoglobin and Myoglobin**

Doctor: **Nayef Karadsheh**

Date: **00/9/2016**

Price:

# Hemoglobin & Myoglobin<sup>1a</sup>

## - Objectives

- Structure-Function relationships in proteins
- Hb - an allosteric protein
- Hb - 4 chains and of two kinds
- Why fetuses have distinctive Hb "Hb F"
- Concept of Molecular Diseases

## A. Hb & Mb

(1) Normal structure  
of Hb & Mb

(2) Effect of  
•  $P_{O_2}$ , pH & Temp.  
- Mechanism of cooperativity

(3) Effect of 2,3-BPG

(4)  $CO_2$  transport & Mechanism  
Bohr effect

(5) Abnormal Hb

(6) Thalassemia

(7) Hb derivatives

B. Heme metabolism

C. Iron metabolism

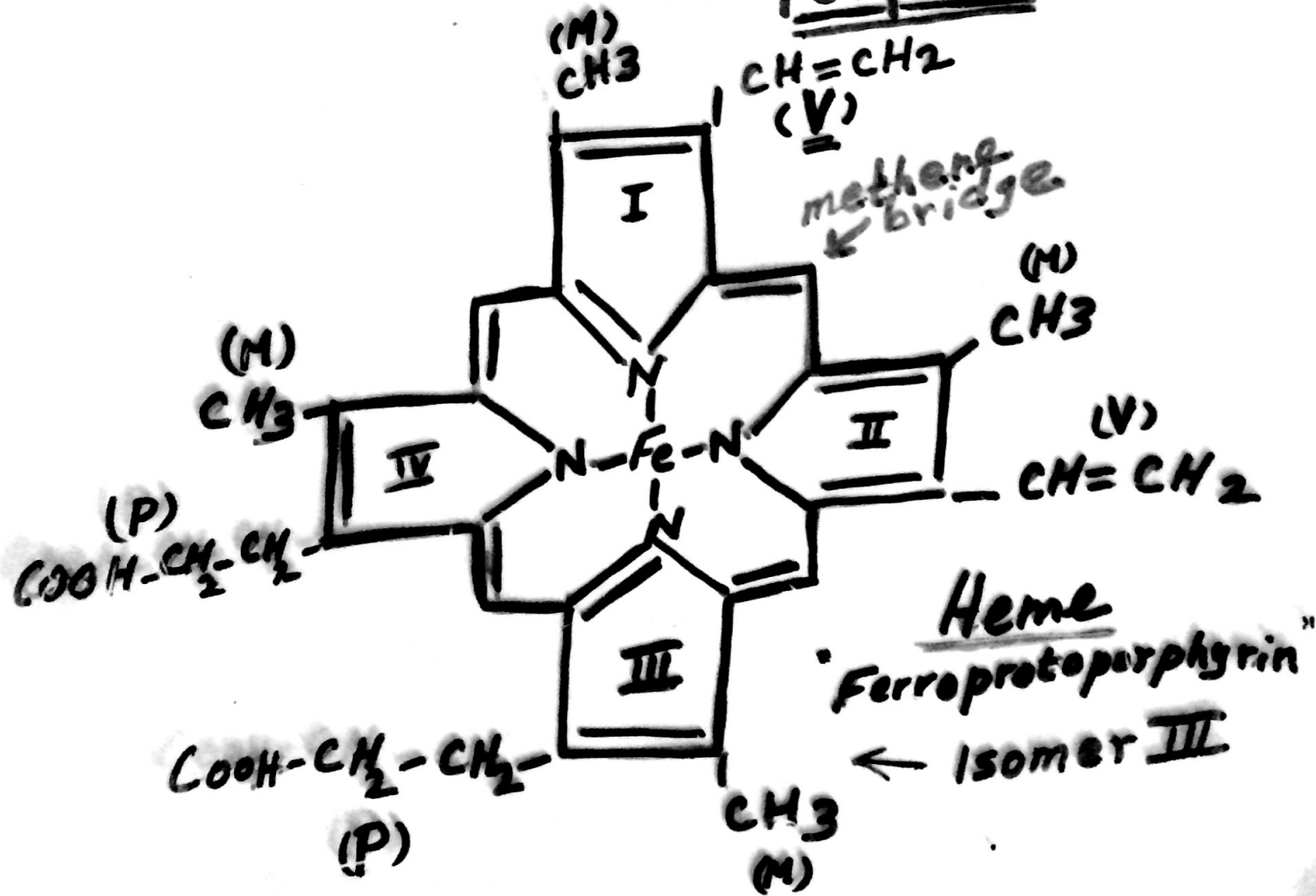
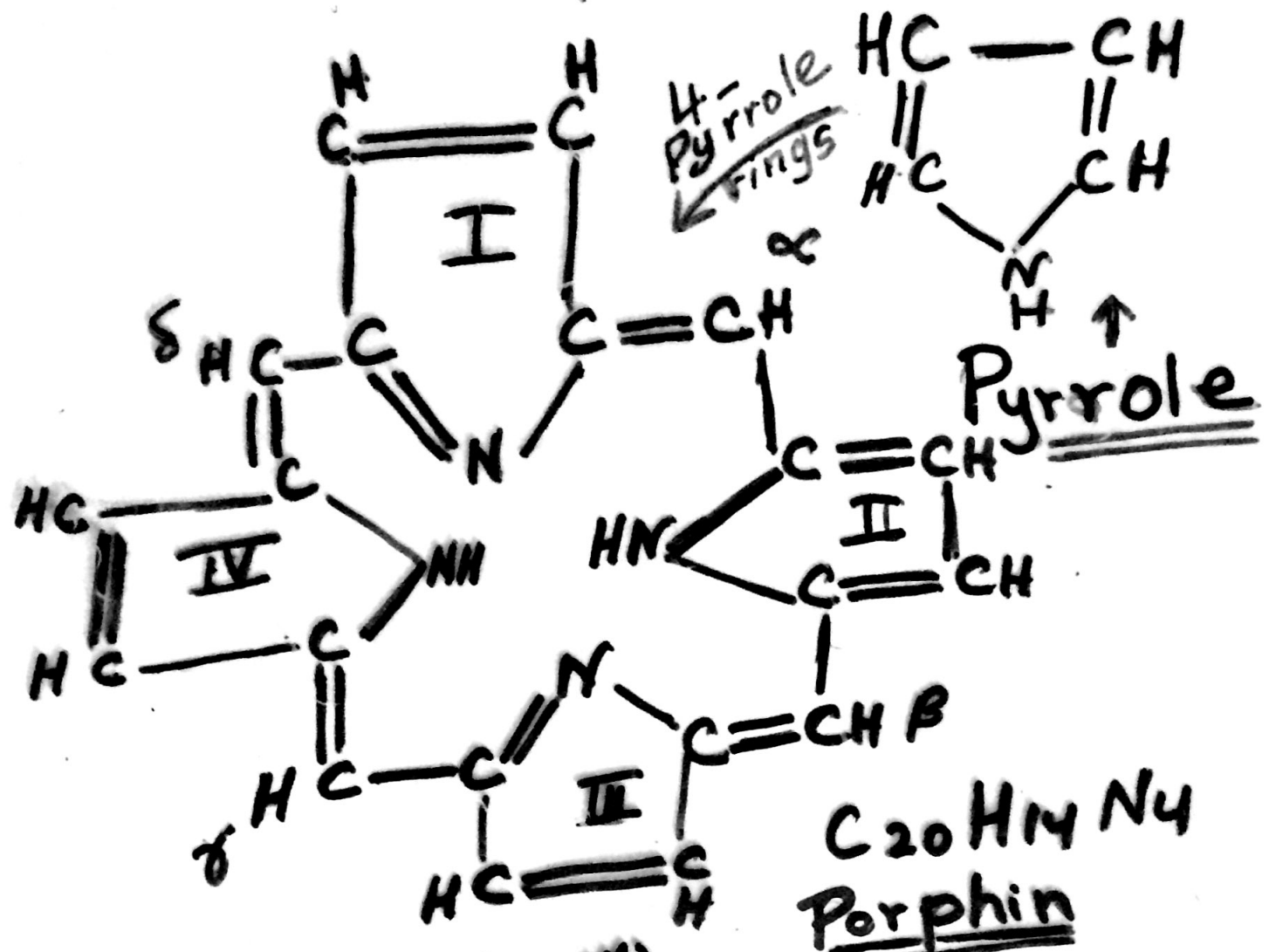
D. Metabolism in mature  
rbc & Genetic disorders

# - Globular Heme proteins

- Prosthetic group - HEME
- Apoprotein - provides an environment of three dimensional structure that dictate the role of heme
  - Reversible binding of  $O_2$  as in Hb and Mb
  - Electron carrier as in cytochromes
  - breakdown of  $H_2O_2$  as in catalase
  - others

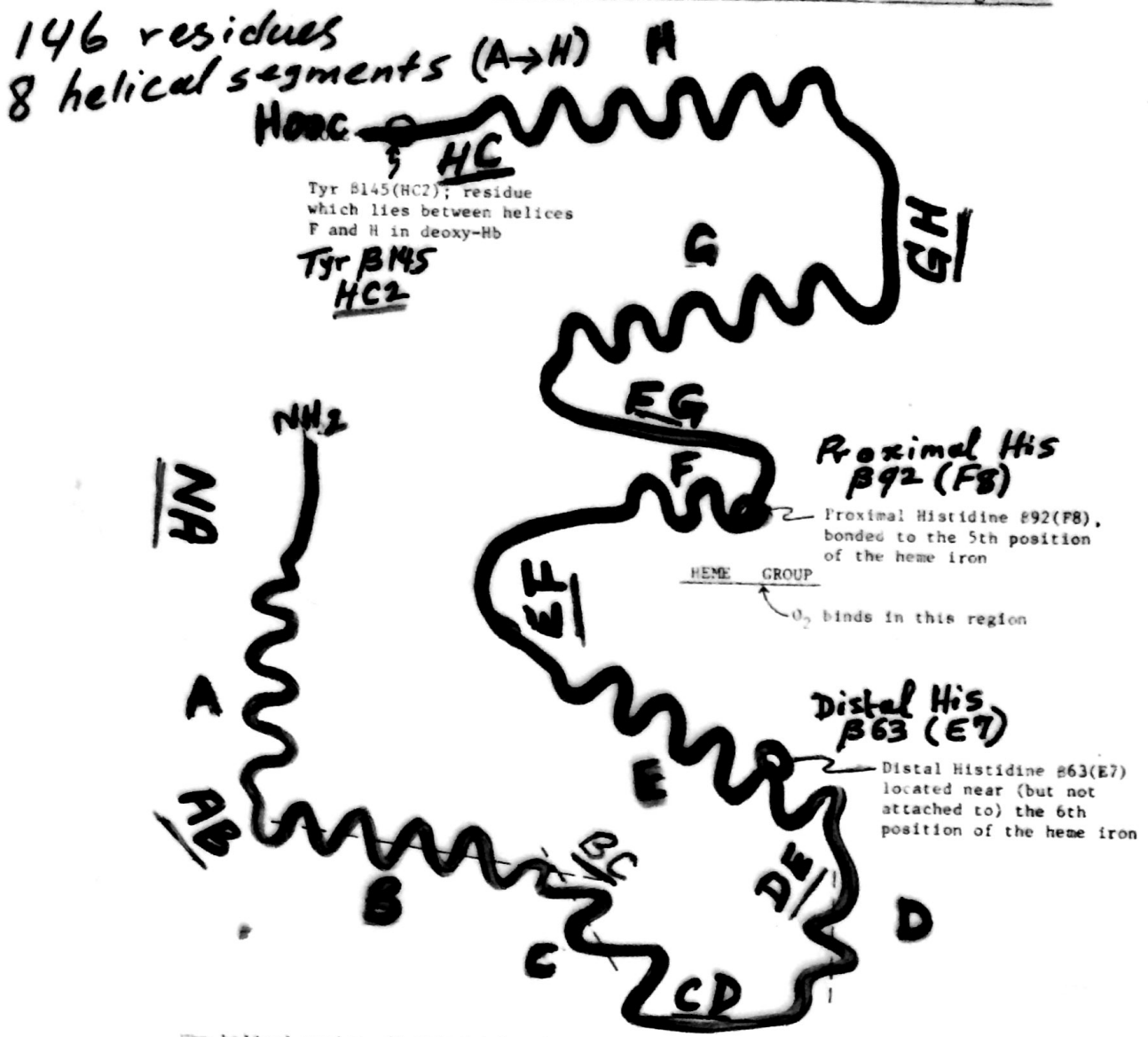


# HEME STRUCTURE:



# SEC. STRUCTURE OF $\beta$ -chain of Hb

Figure 81. Secondary Structure of the  $\beta$ -Chain of Human Hemoglobin



The helical regions (labeled A-H, after Kendrew), N- and C-termini, and the histidines located near the heme group are indicated. The axes of the B, C, and D helices are indicated by dashed lines.

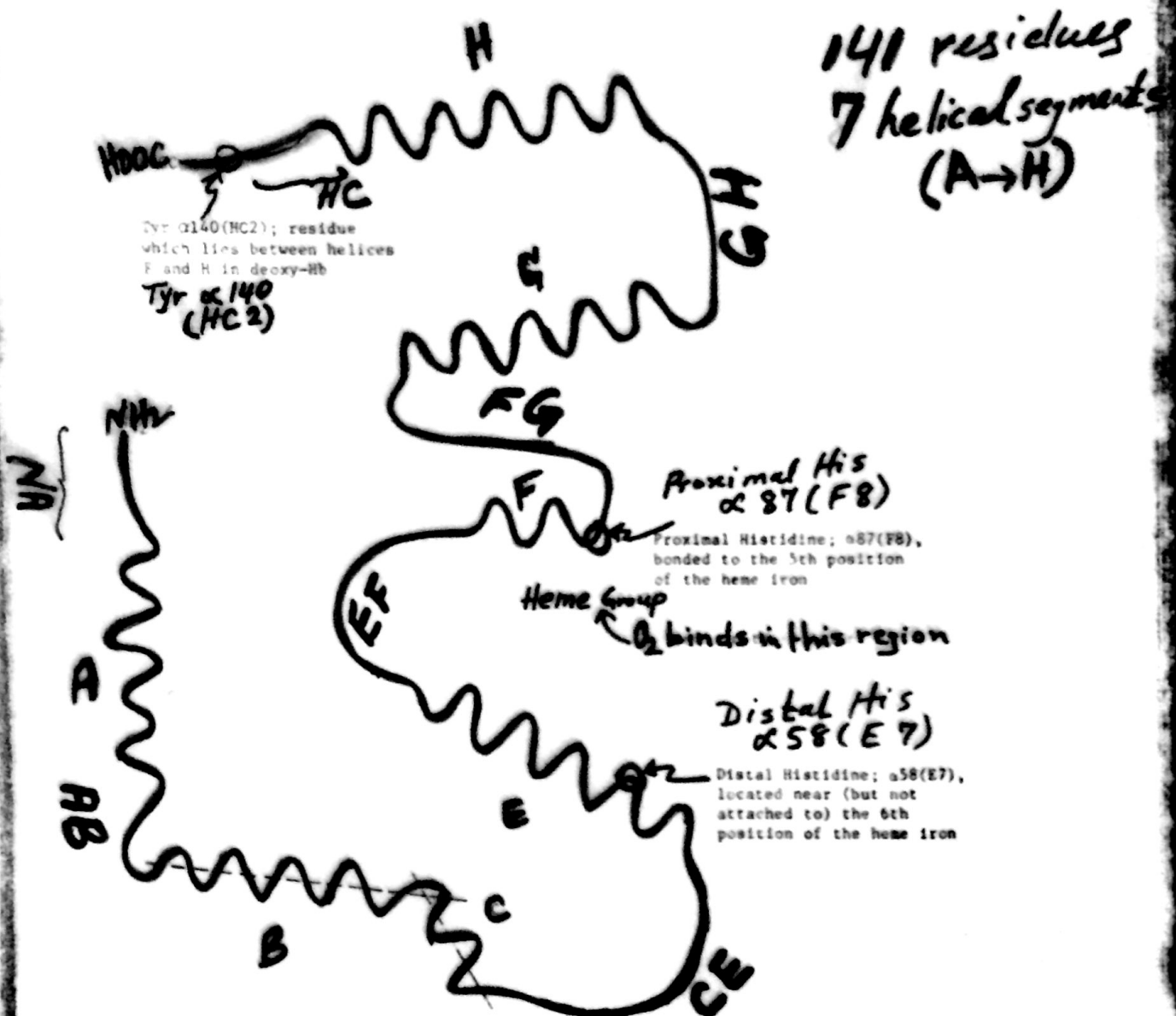
The  $\alpha$ -helical regions are terminated by

- 1- Presence of Proline
- or 2-  $\beta$ -bends and loops stabilized by H-bonds and ionic bonds

Electrostatic Interactions or salt bridges

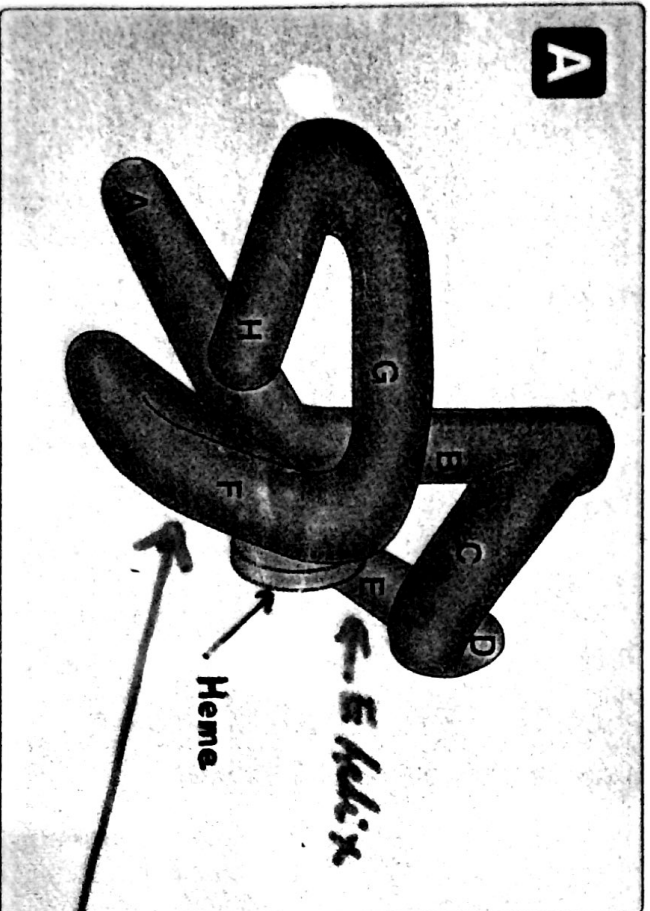
# Sec. Structure of $\alpha$ -chain of Hb <sup>40</sup>

Figure 80. Secondary Structure of the  $\alpha$ -Chain of Human Hemoglobin

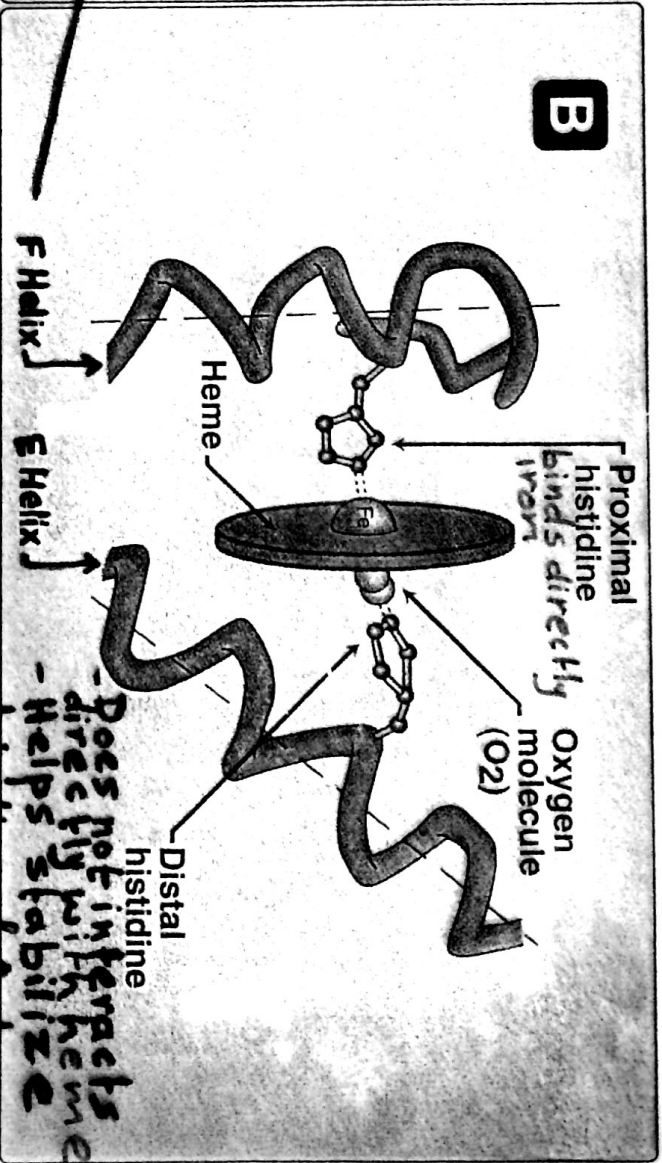


The helical regions (labeled A-H, after Kendrew), N- and C-termini, and the histidines located near the heme group are indicated. The axes of the B and C helices are indicated by dashed lines.

# Binding sites of heme



Prosthetic group  
Apoprotein  
Holo protein



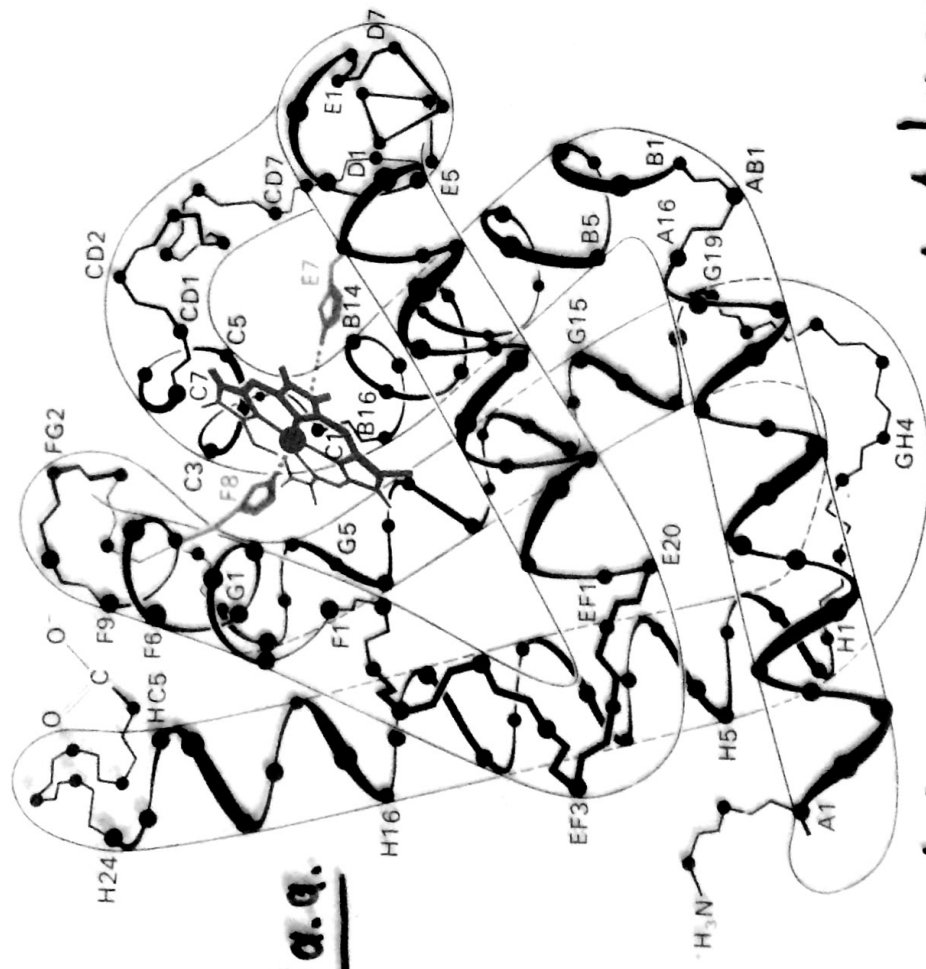
Proximal His  
Distal His  
Heme Pocket

The heme pocket or crevice is lined with non-polar amino acids [except two His] which stabilize hydrophobic heme and permits reversible binding of O<sub>2</sub>. Loss of electrons by Fe is RARE.

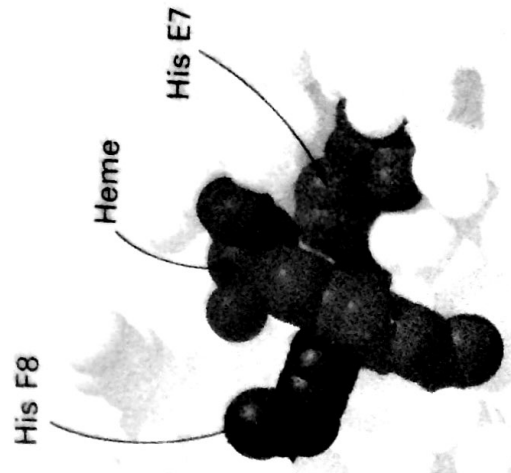
# Tertiary Structure Myoglobin, Mb :-

5

- Mb. is compact 45 X 35 X 25 Å
- ~75% helical structure (8-helical segments)



153 a.a.

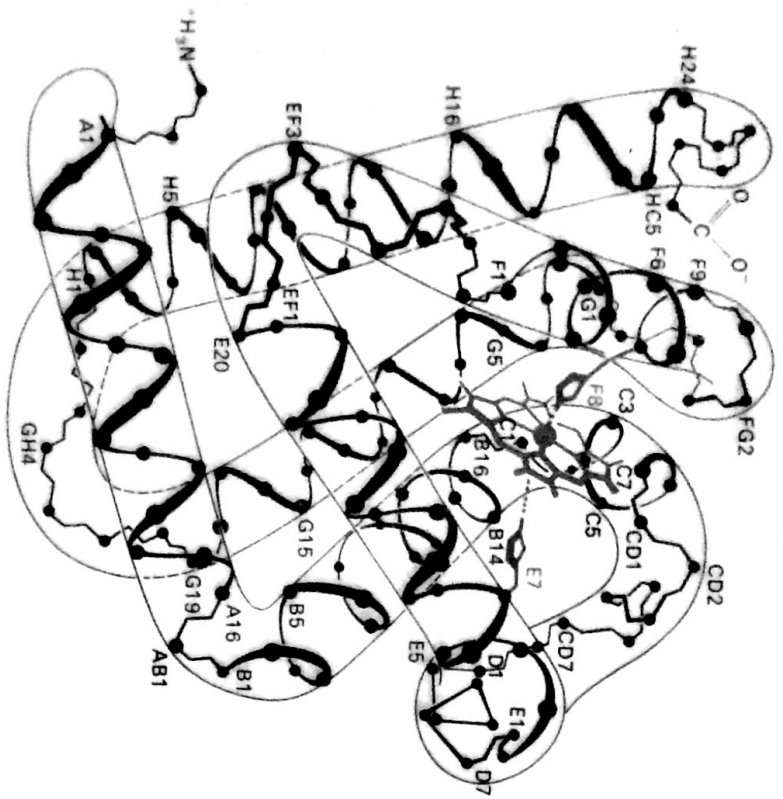


non-polar residues

- 4 helices are terminated by proline
- Interior consist of hydrophobic residue except for prox. 4 dis. his

Figure 7-4, page 149; Figure 7-6, page 150

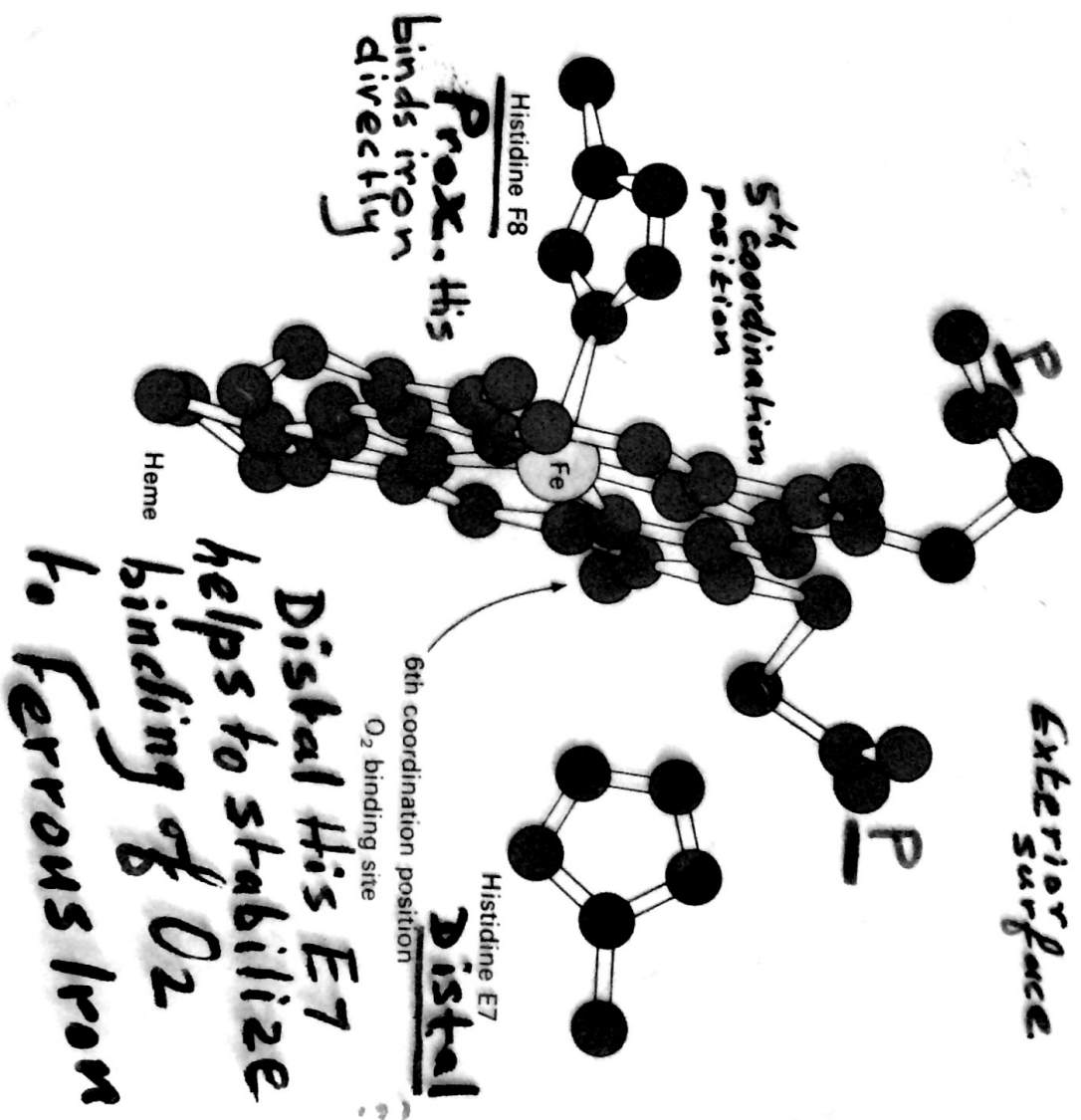
# The O<sub>2</sub>-binding site



## Tertiary Structure

Figures 7-5 and 7-8

Stryer: Biochemistry, Third Edition  
W. H. Freeman and Company





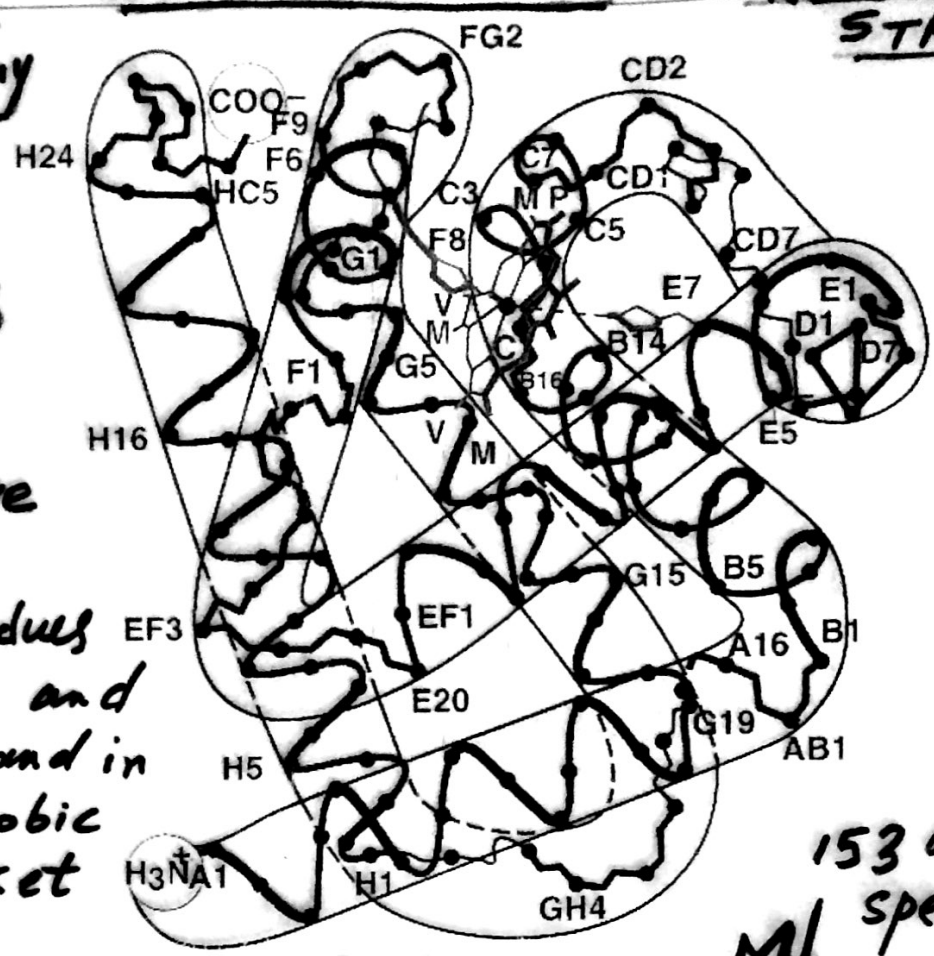
7) The Pri, Sec. & Ter. Structures of Mb & Hb chains  
CLOSE RESEMBLANCE in THREE-DIMENSIONAL STRUCTURE

- 83 Invariant residues in many Mb

- 15 Invariant residues are similar to Hb

• Many of the changes are conservative

Invariant residues include prox. and distal His and in the hydrophobic heme pocket

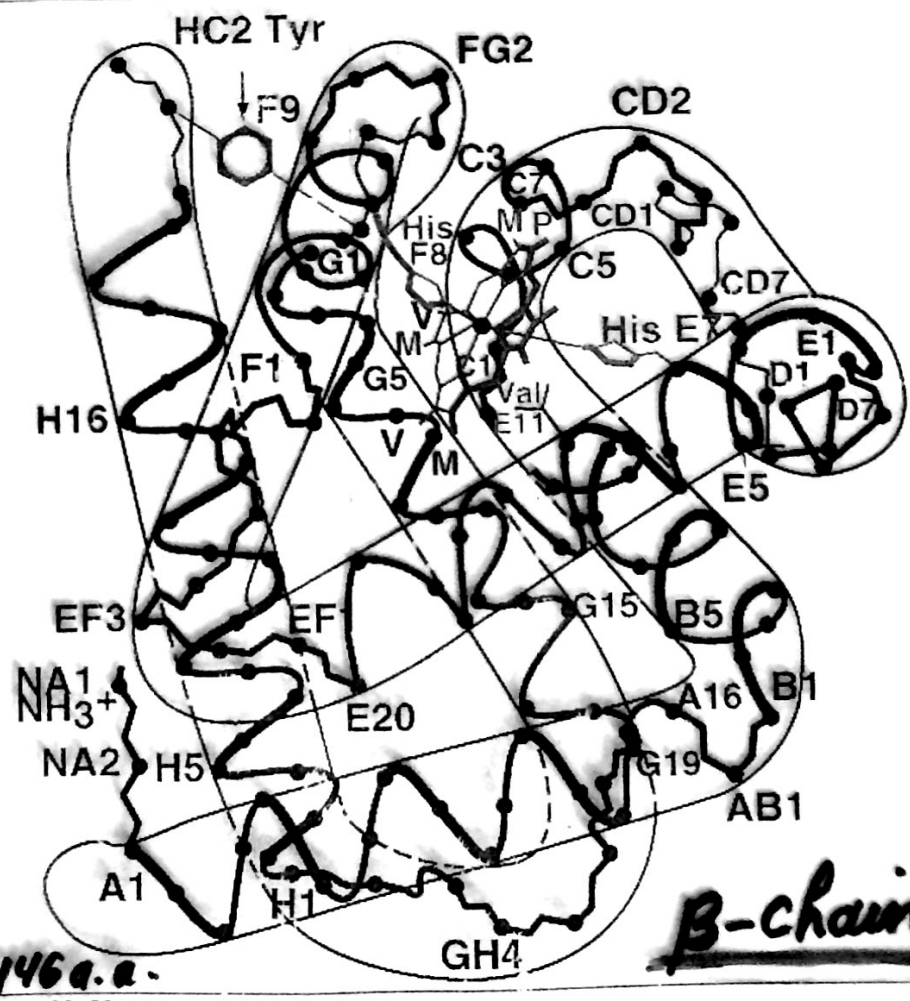


153 a.a. in all Mb species

(a)

153 a.a.

Mb



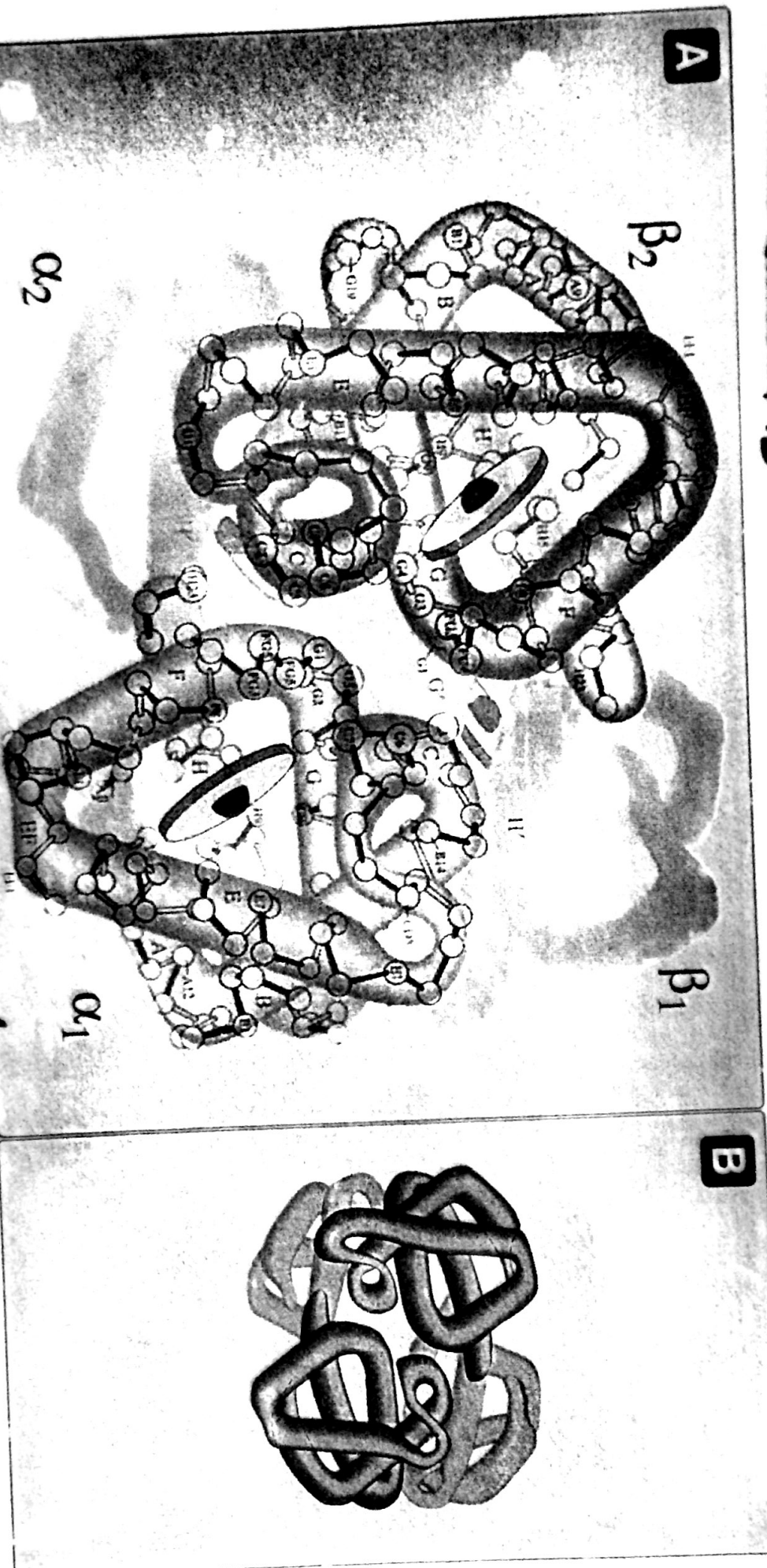
146 a.a.

beta-chain

Figure: 03\_33 Secondary and tertiary structure characteristics of chains of hemoglobin.

# Quaternary Structure of Hb

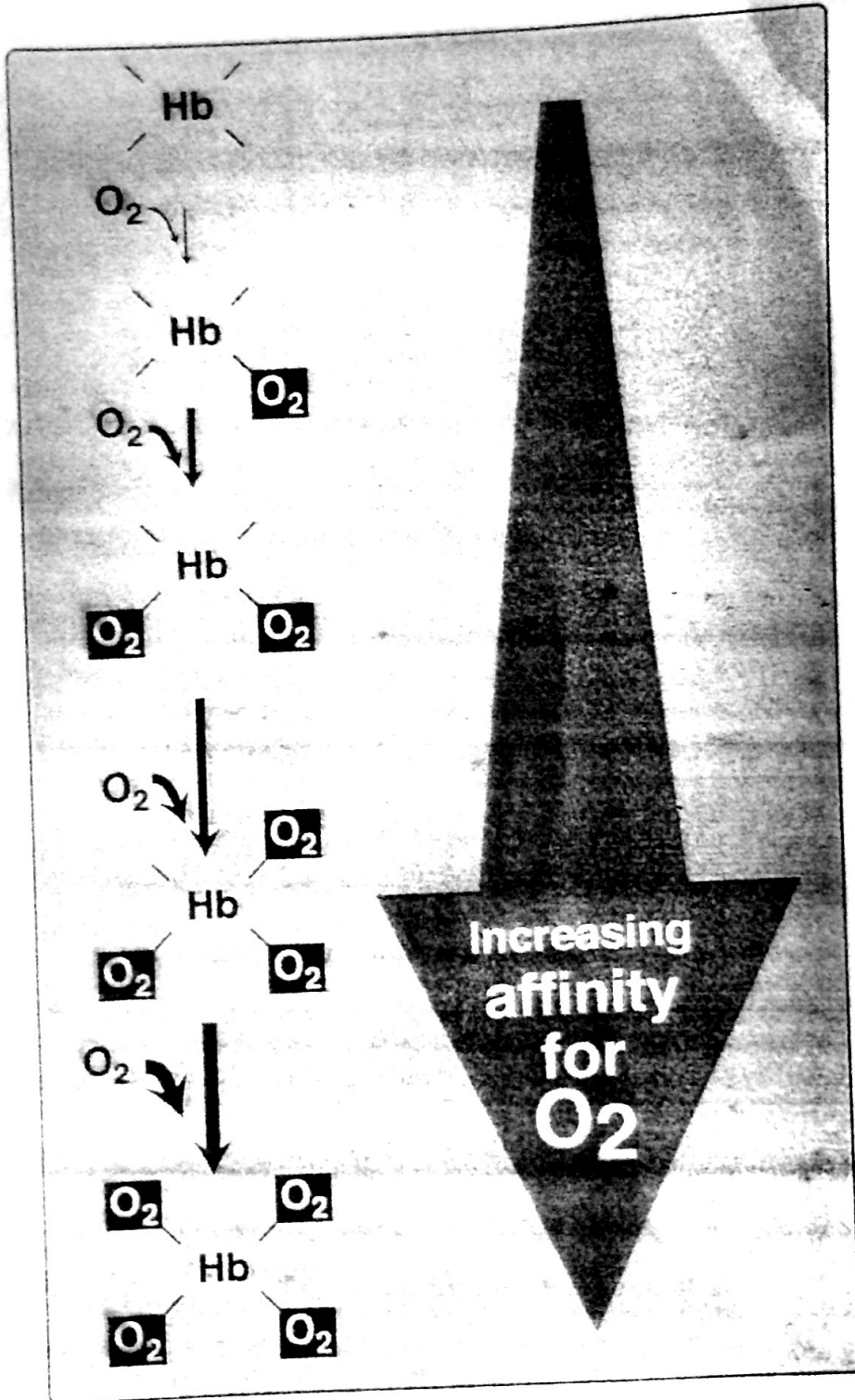
Structure and function of Hb tetramer is more complex than Mb



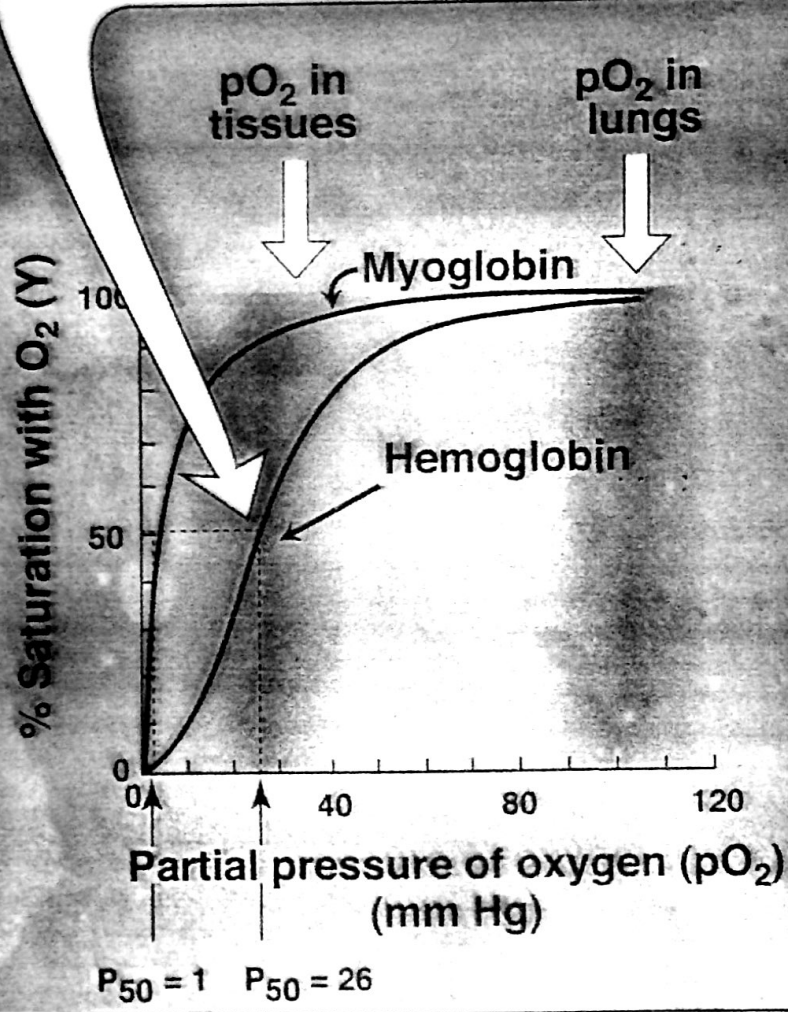
- Hb can transport H<sup>+</sup>, CO<sub>2</sub> and 4 O<sub>2</sub>
- O<sub>2</sub> binding to Hb is regulated by allosteric effectors while Mb is not



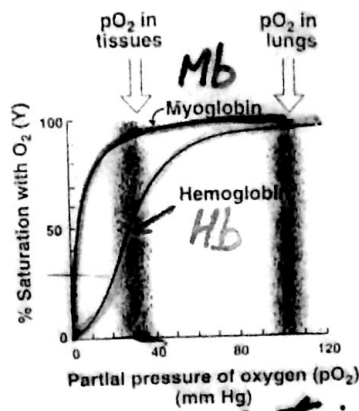
Hb binds successive  $O_2$  molecules with increasing affinity



The oxygen-dissociation curve for Hb is steepest at the oxygen concentrations that occur in the tissues. This permits oxygen delivery to respond to small changes in  $pO_2$ .



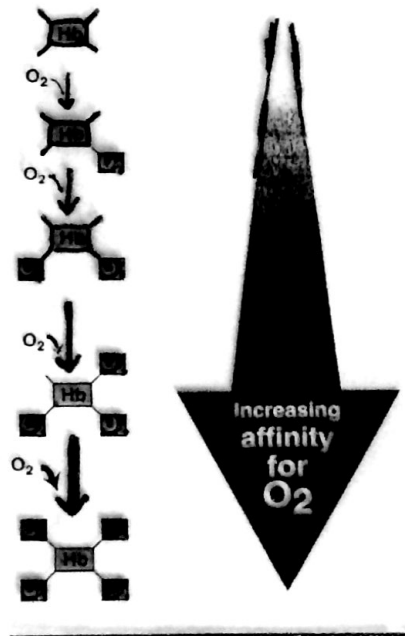
# Binding of oxygen to myoglobin and hemoglobin:-



$P_{50} = \begin{cases} \nearrow \text{Mb} = 1 \\ \searrow \text{Hb} = 26 \end{cases}$

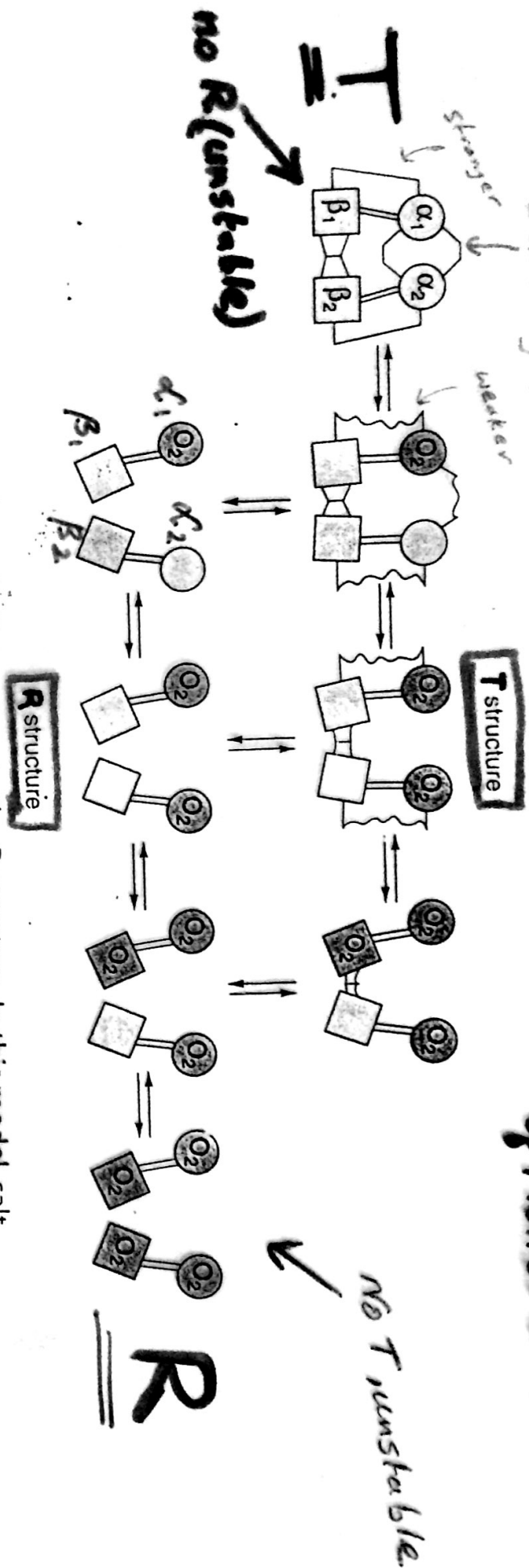
- $O_2$ -dissociation curve for Hb & Mb
- Steepest at  $[O_2]$  in tissue which allow  $O_2$  delivery to respond to small changes in  $P_{O_2}$

-  $O_2$  binds cooperatively to Hemoglobin:-



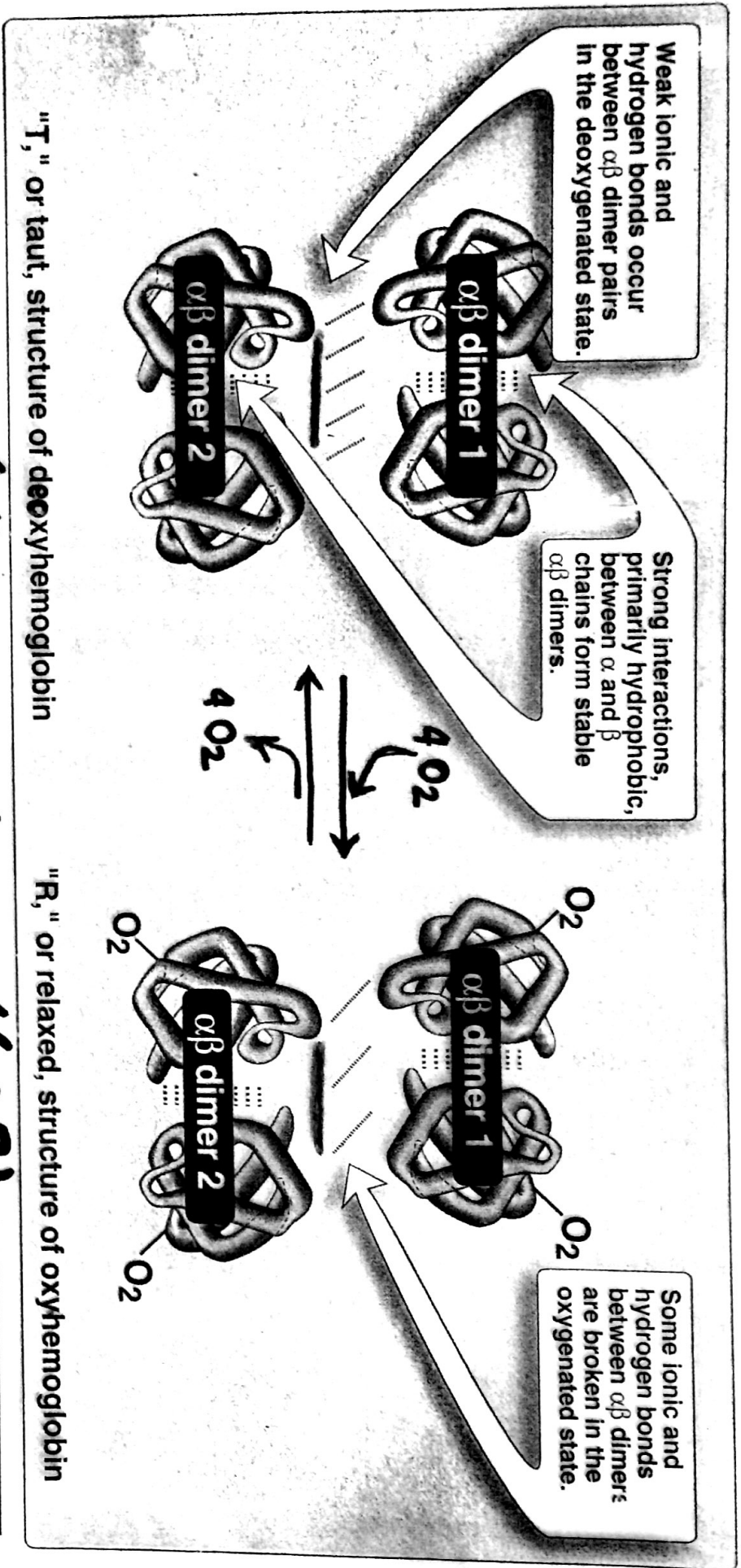
Hb. binds  $O_2$  with increasing affinity

# Model for the transition from T to R [Concerted Model] of Monod et al



**Figure 6-8.** Transition from the T structure to the R structure. In this model, salt bridges (thin lines) linking the subunits in the T structure break progressively as oxygen is added, and even those salt bridges that have not yet ruptured are progressively weakened (wavy lines). The transition from T to R does not take place after a fixed number of oxygen molecules have been bound but becomes more probable as each successive oxygen binds. The transition between the two structures is influenced by protons, carbon dioxide, chloride, and BPG; the higher their concentration, the more oxygen must be bound to trigger the transition. Fully oxygenated molecules in the T structure and fully deoxygenated molecules in the R structure are not shown because they are unstable. (Modified and redrawn, with permission, from Perutz MF: Hemoglobin structure and respiratory transport. Sci Am [Dec] 1970;223:92.)

# Transitional or Conformational Change Upon Oxygenation



"T", or taut, structure of deoxyhemoglobin

"R", or relaxed, structure of oxyhemoglobin

- Two identical dimers ( $\alpha\beta$ )<sub>1</sub> and ( $\alpha\beta$ )<sub>2</sub>
- The two polypeptides in each are held mainly by hydrophobic interactions
  - hydrophobic a.a. in interior and on surface → strong hydrophobic interaction between  $\alpha$  and  $\beta$  in dimer
- The two dimers are held by polar bonds.
- Weaker polar interaction allow movement of dimers



# The O<sub>2</sub>-binding Curve for Hb and Mb

1 kPa = 7.5 torr (mm Hg)

kPa ≡ kilopascal

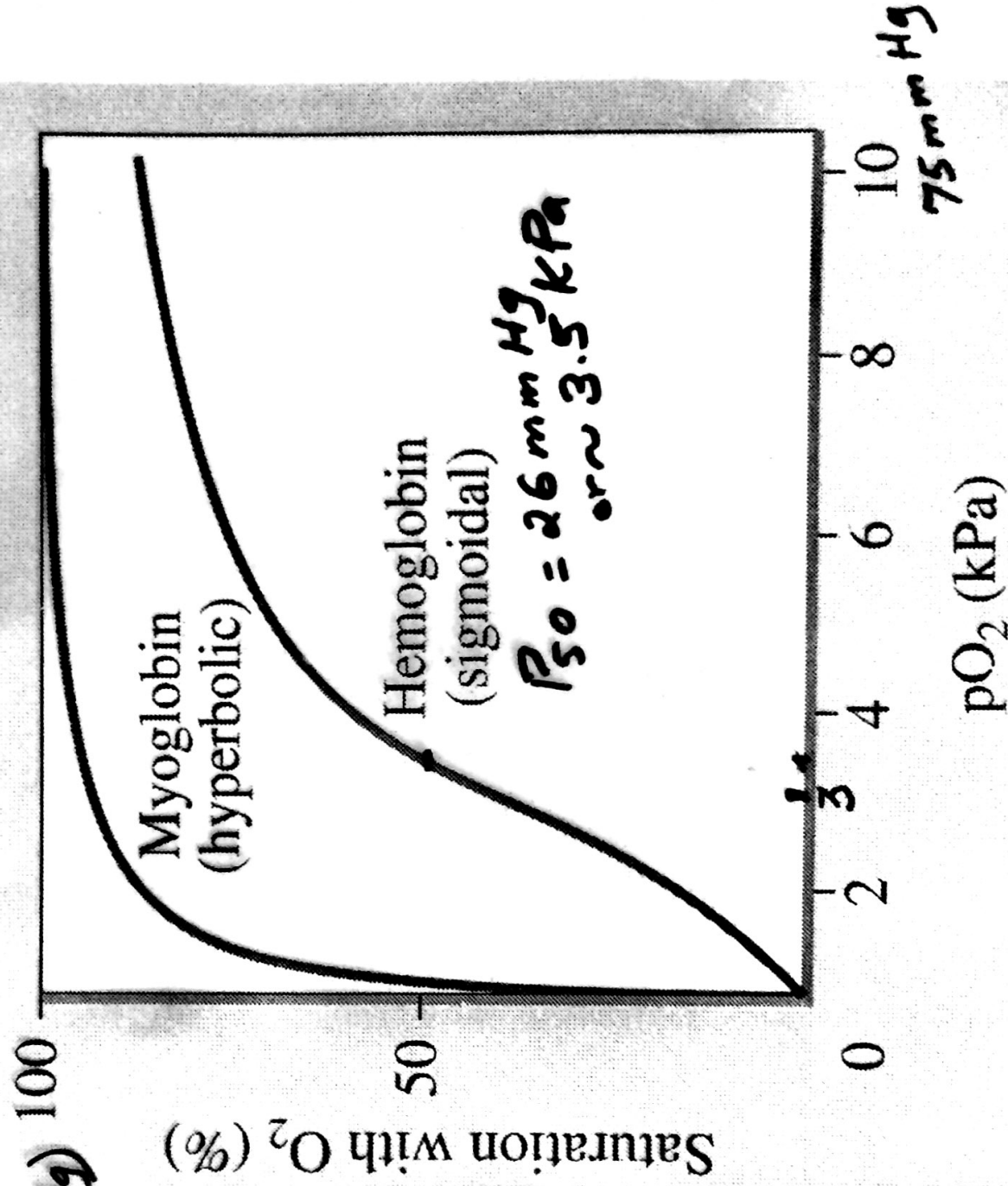


Figure 4-19 Concepts in Biochemistry, 3/e  
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